



**Equations:**

**Electrostatic Force:**  $F_e = \frac{k_e \cdot q_1 \cdot q_2}{d^2}$

**Potential Difference:**  $V = \frac{PE}{q}$   
(Energy per Charge)

**Electric Field:**  $E = \frac{F_e}{q}$

**Data:**

**Electron Charge** =  $-1.60 \times 10^{-19} \text{ C}$

**Proton Charge** =  $+1.60 \times 10^{-19} \text{ C}$

$k_e = 9.0 \times 10^9 \text{ N}\cdot\text{m}^2 / \text{C}^2$

**Electron Mass** =  $9.11 \times 10^{-31} \text{ kg}$

**Proton Mass** =  $1.67 \times 10^{-27} \text{ kg}$

<u>Name</u>	<u>Symbol</u>	<u>Unit</u>	<u>Notes</u>
Electrical Force	$F_e$	Newton	
Charge	$q$	Coulomb microCoulomb ( $\mu\text{C} = 10^{-6} \text{ C}$ )	
Distance	$d$	meters	
Potential Difference	$V$	Volt (Joule/Coulomb)	
Potential Energy	$PE$	Joule	
Kinetic Energy	$KE$	Joule	
Electric Field	$E$	Newtons/Coulomb	

**Helpful Equations:**

$KE = \frac{1}{2} m \cdot v^2$

$W = m \cdot g$

### Electron & Proton Data

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